The CO₂ system in the Mediterranean Sea: a basin wide perspective

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Abstract. The Mediterranean Sea (MedSea) is considered a “laboratory basin” being an ocean in miniature, suffering dramatic changes in its oceanographic and biogeochemical conditions derived from natural and anthropogenic forces. Moreover, the MedSea is prone to absorb and store anthropogenic carbon due to the particular CO₂ chemistry and the active overturning circulation. Despite this, water column CO₂ measurements covering the whole basin are scarce. This work aims to be a base-line for future studies about the CO₂ system space-time variability in the MedSea combining historic and modern CO₂ cruises in the whole area. Here we provide an extensive vertical and longitudinal description of the CO₂ system variables (total alkalinity – TA, dissolved inorganic carbon – DIC and pH) along an East-West transect and across the Sardinia-Sicily passage in the MedSea from two oceanographic cruises conducted in 2011 measuring CO₂ variables in a coordinated fashion, the RV Meteor M84/3 and the RV Urania EuroFleets 11, respectively. In this sense, we provide full-depth and length CO₂ distributions across the MedSea, and property-property plots showing in each sub-basin post-Eastern Mediterranean Transient (EMT) situation with regard to TA, DIC and pH.

The over-determined CO₂ system in 2011 allowed performing the first internal consistency analysis for the particularly warm, high salinity and alkalinity MedSea waters. The CO₂ constants by Mehrbach et al. (1973) refitted by Dickson and Millero (1987) are recommended. The sensitivity of the CO₂ system to the atmospheric CO₂ increase, DIC and/or TA changes is evaluated by means of the Revelle and buffer factors.

1 Introduction

The Mediterranean Sea (MedSea hereafter) is a particular system with respect to its physical and biogeochemical oceanography (e.g. Tanhua et al., 2013a). The MedSea is a land-locked sea exporting warm and salty intermediate water into the Atlantic Ocean through the shallow and narrow Strait of Gibraltar affecting the global thermohaline circulation (e.g. McCartney and Mauritzen, 2001). Apart from the shallow open thermohaline cell encompassing the whole Mediterranean basin, two closed deep overturning cells are active in the western and eastern basins, respectively. Additionally, a complex upper layer circulation is present including significant permanent and quasi-permanent eddies.

One of the most relevant biogeochemical characteristics of the MedSea, especially the eastern basin, is the anomalous high Nitrogen to Phosphorous ratio (e.g. Ribera d’Alcalà et al., 2003; Krom et al., 2005; Pujo-Pay et al., 2011). This fact along with the oligotrophic gradient from west to east could be attributed to the difference in the allochthonous nutrient sources in terms of quantity and quality (Ribera d’Alcalà et al., 2003; Ludwig et al., 2010; Schroeder et al., 2010; Krom et al., 2010) and also to the specific hydrodynamic features of the MedSea (Robinson et al., 2001; Millot and Taupier-Letage, 2005; Huertas et al., 2012).

With a limited exchange with the ocean and its specific circulation, the MedSea is considered a “laboratory basin” (e.g. Robinson and Golnaraghi, 1994; Bergamasco and Malanotte-Rizoli, 2010) where processes occurring on a global scale can be approximated to smaller time and space...
scales. This fact mostly mentioned for general circulation studies can also be applied for biogeochemical studies, eutrophication, hypoxia, acidification, etc.

The MedSea represents only 0.8 % of the global oceanic surface and despite the general sparseness of water column CO$_2$ measurements (Álvarez, 2012) it has been identified as an important anthropogenic carbon storage where the column inventory is much higher than in the Atlantic or Pacific oceans (Schneider et al., 2010; Lee et al., 2011). The reasons for this are the intrinsic physico-chemical characteristics of the MedSea where warm and high alkalinity waters, thus low Revelle factor, are prone to absorb CO$_2$ from the atmosphere and be transported to the interior by the active overturning circulation. With regard to interior CO$_2$ data, the published literature discussed specific cruises, in the eastern MedSea sub-basins (e.g. Schneider et al., 2010; Luchetta et al., 2010; Krasakopoulou et al., 2011) and the western MedSea (e.g. Brunet et al., 1984;Claustre et al., 2001; Copin-Montegut, 1993; Millero et al., 1979). Cruises covering the eastern and western MedSea are few: the TRANSMED cruise in 2007 (Rivaro et al., 2010), BOUM in 2008 (Touratier and Goyet, 2012), ours in 2011 and the not yet published MedSeA 2013 cruise (medseaoceanacruise.wordpress.com).

This paper is a contribution to the Special Issue “Physical, chemical and biological oceanography of the Mediterranean Sea”. In particular, it focuses on CO$_2$ system analysis performed on board RV Meteor (M84/3 cruise) and RV Urania (EuroFleets 11 cruise), both sampled the MedSea in April 2011 (Tanhua et al., 2013a) (Fig. 1). Besides characterising the particular chemistry of the CO$_2$ in the MedSea, we will present and discuss vertical distributions and property-property plots in order to present a quasy-synoptic picture of the post-EMT oceanographic situation with regard to the CO$_2$ system. For those not familiar, briefly, the EMT was a dramatic event changing the hydrography and the large-scale circulation of deep waters in the eastern MedSea, which changed the formation area from the Adriatic to the Aegean Sea in the 1990s (e.g. Roether et al., 1996, 2007).

2 Data and methods

2.1 Meteor M84/3 cruise

During the Meteor M84/3 cruise in April 2011 (Fig. 1) (Tanhua et al., 2013a, b) Dissolved Inorganic Carbon (DIC), pH and Total Alkalinity (TA) were measured in all stations and depths in order to have an over-determined CO$_2$ system. Details about the sampling, analysis and quality control are given in Tanhua et al. (2013b). The bottle data is available at the Carbon Dioxide Information Analysis Center (CDIAC) (Tanhua et al., 2012).

The DIC content was determined using a SOMMA instrument. Samples were collected in borosilicate bottles according to standard operation protocol. The precision of the analysis was determined to $\pm 0.6 \mu$mol kg$^{-1}$ by titration of several bottles filled from the same Niskin bottle. The accuracy was determined to be $2.5 \mu$mol kg$^{-1}$ by analysing a total of 42 bottles of certified reference material (CRM, Andrew Dickson, Scripps, CA, USA, batch 108); the DIC of this batch is certified at 2022.70 $\pm 0.7 \mu$mol kg$^{-1}$. Measurements of the CRMs were also used to daily correct the temporal drift in the coulometer cell; this correction was never larger than 3 $\mu$mol kg$^{-1}$.

pH was measured in cylindrical optical glass 10-cm path-length cells with a spectrophotometric procedure (Clayton and Byrne, 1993). pH is reported in the total scale at 25 $^\circ$C (pH25T). The precision was estimated to $\pm 0.0012$ by measuring replicates from the same Niskin bottle. Regarding the accuracy, the theoretical pH25T value for the 108 CRM batch using the dissociation constants from Mehrbach et al. (1973) refitted by Dickson and Millero (1987) is 7.8782. Replicate pH measurements along the cruise on CRMs are lower than
the theoretical value by 0.096 0.0012 pH units. We will discuss this point in Sect. 3.2.

Samples for TA determination were collected in 600 mL borosilicate bottles and were analysed within one day after sampling. The TA samples were analysed following the double end point potentiometric technique by Pérez and Fraga (1987) and Pérez et al. (2000). Measurements of CRM were performed in order to control the accuracy of the TA measurements; the pH as measured by the electrode was corrected to obtain the closest mean TA to the CRM; the pH measurements; the pH as measured by the electrode was corrected to obtain the closest mean TA to the CRM; the pH measurements; the pH as measured by the electrode was corrected to obtain the closest mean TA to the CRM; the pH measurements; the pH as measured by the electrode was corrected to obtain the closest mean TA to the CRM; the pH measurements; the pH as measured by the electrode was corrected to obtain the closest mean TA to the CRM; the pH measurements; the pH as measured by the electrode was corrected to obtain the closest mean TA to the CRM; the pH measurements; the pH as measured by the electrode was corrected to obtain the closest mean TA to the CRM; the pH measurements; the pH as measured by the electrode was corrected to obtain the closest mean TA to the CRM; the pH measurements; the pH as measured by the electrode was corrected to obtain the closest mean TA to the CRM; 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the pH measurements; the pH as measured by the electrode was corrected to obtain the closest mean TA to the CRM; the pH measurements; the pH as measured by the electrode was corrected to obtain the closest mean TA to the CRM; the pH measurements; the pH as measured by the electrode was corrected to...
Table 1. Mean values and standard deviation of the residuals (measured minus calculated) for alkalinity (TA), Dissolved Inorganic Carbon (DIC) and pH at the total scale and referred to 25°C (pH25T) calculated with different thermodynamic constants for CO2 according to the CO2SYS programme options. The sulphate constant from Dickson (1990) and total borate concentration from Kho et al. (1977) were used in all cases.

<table>
<thead>
<tr>
<th>Constants</th>
<th>TA</th>
<th>DIC</th>
<th>pH25T</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>-16.1 ± 4.0</td>
<td>14.7 ± 3.6</td>
<td>0.024 ± 0.005</td>
</tr>
<tr>
<td>2</td>
<td>-11.1 ± 3.7</td>
<td>10.2 ± 3.4</td>
<td>0.016 ± 0.005</td>
</tr>
<tr>
<td>3</td>
<td>-5.0 ± 3.5</td>
<td>4.6 ± 3.2</td>
<td>0.007 ± 0.005</td>
</tr>
<tr>
<td>4</td>
<td>0.1 ± 3.3</td>
<td>-0.1 ± 3.0</td>
<td>-0.0003 ± 0.0005</td>
</tr>
<tr>
<td>5</td>
<td>-3.7 ± 3.5</td>
<td>3.4 ± 3.2</td>
<td>0.005 ± 0.005</td>
</tr>
<tr>
<td>6</td>
<td>6.3 ± 3.2</td>
<td>-5.9 ± 3.0</td>
<td>-0.009 ± 0.005</td>
</tr>
<tr>
<td>7</td>
<td>-3.6 ± 3.5</td>
<td>3.3 ± 3.2</td>
<td>0.005 ± 0.005</td>
</tr>
<tr>
<td>8</td>
<td>17.3 ± 3.2</td>
<td>-16.1 ± 3.0</td>
<td>-0.027 ± 0.005</td>
</tr>
<tr>
<td>9</td>
<td>0.6 ± 3.3</td>
<td>-0.6 ± 3.0</td>
<td>-0.001 ± 0.005</td>
</tr>
<tr>
<td>10</td>
<td>-4.3 ± 3.6</td>
<td>3.9 ± 3.3</td>
<td>0.006 ± 0.005</td>
</tr>
<tr>
<td>11</td>
<td>-10.2 ± 4.4</td>
<td>9.3 ± 4.0</td>
<td>0.015 ± 0.006</td>
</tr>
<tr>
<td>12</td>
<td>2.8 ± 3.3</td>
<td>-2.6 ± 3.1</td>
<td>-0.004 ± 0.005</td>
</tr>
<tr>
<td>13</td>
<td>1.5 ± 3.4</td>
<td>-1.4 ± 3.1</td>
<td>-0.002 ± 0.005</td>
</tr>
</tbody>
</table>

(1) Roy (1993); (2) Goyet and Poisson (1989); (3) Hansson (1973a, b) refit by Dickson and Millero (1987); (4) Mehrbach et al. (1973) refit by Dickson and Millero (1987); (5) Hansson (1973a, b) and Mehrbach et al. (1973) refit by Dickson and Millero (1987); (6) GEOSECS choice (Takahashi et al., 1982); (7) Peng et al. (1987); (8) Cai and Wang (1998); (9) Lueker et al. (2000); (10) Mojica-Prieto and Millero (2002); (11) Millero et al. (2002); (12) Millero et al. (2006); (14) Millero (2010).

Despite the lack of surface or column partial pressure CO2 (pCO2) measurements, this study presents high-quality DIC, TA and pH data covering the whole MedSea that can be used to perform the first internal consistency analysis in this area for CO2 measurements.

By using the consensus programme for seawater CO2 calculations CO2SYS, configured for ®MATLAB by van Heuven et al. (2011), we tested the internal consistency of our measurements for the various CO2 constants using all options, except that for pure water only (option 8). We also changed the sulfate constants (Dickson, 1990; Kho, 1977) and the parameterization of borate as a function of salinity (Uppström, 1974; Lee et al., 2010). The inorganic nutrients concentrations were also taken into account.

The option yielding minimum differences between the calculated and measured variable for the three combinations (TA-pH, TA-DIC and DIC-pH) (Table 1 and Fig. 2) is obtained by using option 4 [CO2 constants from Mehrbach et al., 1973; refitted by Dickson and Millero, 1987], the sulphate constant from Dickson (1990) and the parameterization of borate from Uppström (1974). Using this borate function and the sulphuric constant from Kho (1977) yields practically the same results (not shown). Using option 13 (Millero et al., 2006), the more recent borate function by Lee et al. (2010) and either sulphate constant (Kho, 1977 or Dickson, 1990) low residuals are also obtained (not shown).

By using our preferred set of constants (see above) our pH, TA and DIC measurements along the M84/3 cruise are internally consistent to −0.0003 ± 0.005 for pH, 0.1 ± 3.3 µmol kg⁻¹ for TA and −0.1 ± 3 µmol kg⁻¹ for DIC. This combination of constants is also the preferred option in global ocean CO2 data synthesis projects such as GLODAP (Key et al., 2004) and CARINA (Key et al., 2010). At the light of these results, the bias of 0.0962 ± 0.0012 in the pH25T measurements obtained after CRM measurements for pH during M84/3 was disregarded.

The effect of uncertainties in pK1 and pK2 using the pH – TA combination is identical to those observed using pH – DIC. As summarized in Lee et al. (1997) when pH and TA (pH and DIC) are used as input parameters, calculated DIC (TA) depends on pK2, which is also dependent upon the DIC/TA ratio; the lower the ratio the higher the sensitivity to pK2. However, only a reliable pK1 is needed to calculate pCO2 from the pH–TA (pH–DIC) combination; in this case the lower the DIC/TA ratio the lower the sensitivity to pK1. Calculated pH from the TA–DIC input combination relies on a good estimation of pK1−pK2 (the K1/K2 ratio). The DIC/TA ratio is an indicative of the concentration of carbonates; the lower the ratio the higher the [CO3]²⁻ concentration, which is also sensitive to pK2. Lacking measured pCO2 data we are not able to check the dependence on K1 in the calculation of pCO2 (pH-DIC or pH-TA inputs) or pH (pCO2-DIC or pCO2-TA inputs).

Having the former information in mind, a closer look at the pH, TA and DIC residuals (measured minus calculated values) reveals some insights and peculiarities about the internal consistency of CO2 measurements in the high TA (low DIC/TA) and salinity MedSea waters. Figure 3 shows the longitudinal distribution of the residuals in the MedSea pointing to a decreasing trend eastwards for TA (Fig. 3a) and on the contrary an increasing trend eastwards for DIC and pH (Fig. 3b and c) which seem to be related with the eastwards salinity increase (Fig. 3 in Tanhua et al., 2013a), DIC/TA ratio decrease (Fig. 10h) or [CO3]²⁻ increase. The pH residuals are a function of pK1−pK2, but mostly dependent on K2, while the TA or DIC residuals are a function of pK2 (Lee et al., 1997; Millero, 2007). The relation between pH and TA residuals is linear (Fig. 4), but two regression lines are detected depending on salinity. Samples with salinity higher than 36.1 have a higher slope (−0.0015 ± 4.5 × 10⁻⁶; r² = 0.9937, n = 725) than fresher ones (−0.0020 ± 2.9 × 10⁻⁵; r² = 0.9945, n = 28). Including measured pH, TA and DIC data from five CARINA cruises (Key et al., 2010) between the Strait of Gibraltar and 24°W and 40°N to 24°N the same tendency is detected (data with salinity < 36.1, slope = −0.0022 ± 8.7 × 10⁻⁶, r² = 0.9919, n = 671; data with salinity ≥ 36.1, slope = −0.0016 ± 6.7 × 10⁻⁶, r² = 0.9977, n = 182). This result points to the need for a revision of the K2 parameterization for waters with high salinity and low DIC/TA ratio, as in the MedSea, as the CO2 chemistry is more sensitive to K2.
Fig. 2. Vertical distribution of the residual (measured minus calculated) values for TA (µmol kg\(^{-1}\)), DIC (µmol kg\(^{-1}\)) and pH25T using option 4 in CO2SYS, see Table 1.

Fig. 3. Same as in Fig. 2 but plotted vs. longitude. The colour code corresponds to the DIC/TA ratio. Two stations to the west of the Strait of Gibraltar were coded as red because DIC/TA ratio > 0.94.
i. Levantine Surface Water (LSW): this water formed by intensive heating and evaporation, has the largest salinity and temperature of the entire MedSea (> 17 °C, > 38.9), clearly seen in the surface layer of the Levantine Basin (Tanhua et al., 2013a). LSW has TA values around 2610 µmol kg⁻¹, DIC around 2270 µmol kg⁻¹ and pH25T around 8.03 (Figs. 5–7).

ii. Atlantic Water (AW): enters through the Strait of Gibraltar and flows eastward, it is mainly detected by a low salinity signal in the upper 50–100 dbars (Fig. 3 in Tanhua et al., 2013a) from the Strait of Gibraltar to the Levantine Basin (36.2–38.9). AW is also characterised by a low TA signal evolving from values < 2560 µmol kg⁻¹ in the western basin to < 2600 µmol kg⁻¹ in the Levantine Basin and Ionian Sea (Fig. 5). The same trend was seen in DIC evolving from 2250 µmol kg⁻¹ in the western to 2270 µmol kg⁻¹ in the Levantine Basin (Fig. 6), pH25T increased from 8 to 8.014 eastwards (Fig. 7).

The MedSea is considered one of the most oligotrophic areas in the world (Azov, 1991) with low phytoplankton biomass and primary production decreasing eastward, confirmed by in situ measurements (Ignatiades et al., 2009; Moutin and Raimbault, 2002; Siokou-Frangou et al., 2010), satellite data (Bosc et al., 2004; Volpe et al., 2007; D’Ortenzio and Ribera d’Alcà, 2009) and modelling studies (Crise et al., 1999; Lazzari et al., 2012). The period of the M84/3 cruise, April, coincides with the end of the winter-spring bloom (D’Ortenzio and Ribera d’Alcà, 2009; Lazzari et al., 2012). Surface Chlorophyll a (Chl a) and primary production data during the M84/3 cruise were presented in Rahav et al. (2013) with a clear westward increasing gradient, and values ranging from 0 to 0.4 µg L⁻¹ for Chl a, and 0.2 to 15 µgC L⁻¹ d⁻¹ for primary production.

The gathered biogeochemical data during the M84/3 cruise showed that the upper 50 dbars are supersaturated in oxygen (AOU < 0) with more negative values in the western basin where higher primary production was reported (Rahav et al., 2013). While in the western basin inorganic nutrients were still available, in the eastern basin they were depleted, especially phosphate (see corresponding figures in Tanhua et al., 2013a). For waters with pressure < 60 dbars, there was no clear relationship between AOU and either DIC, TA or pH25T, except in the western basin where pH25T related with AOU (pH25T = -0.0015 ± 0.00008 · AOU + 7.97 ± 0.001, r² = 0.83 p < 0.001), pointing to a pH modulation by primary production.

iii. Levantine Intermediate Water (LIW): this warm and high salinity water is formed in the Levantine Basin, mainly in the Rhodes Gyre area. During the M84/3 cruise LIW was clearly detected by the salinity maximum around 200 dbars in the eastern basin (Tanhua et al., 2013a), with TA > 2620 µmol kg⁻¹, DIC > 2270 µmol kg⁻¹ and pH25T < 8.014 (Figs. 5–7).
Fig. 5. Sections of TA (μmol kg⁻¹) in the Mediterranean Sea from the Meteor cruise M84/3 in April of 2011. The top right panel is a meridional section from the Adriatic Sea to the Ionian Sea (light gray line on the map) and the lower panel is the zonal section from the coast of Lebanon in the Eastern Mediterranean Sea to through the Strait of Gibraltar (dark gray line on the map). The depth scale and the colour scale are identical all both panels. The top 500 dbars in each section are slightly expanded. No stations are shown in the Atlantic due to the very different TA encountered there as compared to the Mediterranean Sea.

Fig. 6. As in Fig. 5 but for DIC (μmol kg⁻¹).
DIC $> 2310 \mu\text{mol kg}^{-1}$, but no clear signal in pH which varies between 7.96 and 7.98 (Figs. 5, 6 and 7). In the western basin LIW was found around 500 dbar as a maximum in salinity, corresponding to TA ($\approx 2590 \mu\text{mol kg}^{-1}$) (Fig. 5), maximum in DIC ($\approx 2330 \mu\text{mol kg}^{-1}$) (Fig. 6) and minimum pH$_{25T}$ ($\approx 7.9$) (Fig. 7). In the eastern basin, the maximum layer of remineralization, indicated by the oxygen minimum (see Fig. 6 in Tanhua et al., 2013a), was located below LIW while in the western basin, both coincided.

iv. Eastern Mediterranean Deep Water (EMDW): along the M84/3 cruise track in the eastern MedSea three types of EMDW could be distinguished mainly using salinity and AOU (Tanhua et al., 2013a) and tracers such as CFC-12 (Stöven and Tanhua, 2013). Around 1000 dbar in the Levantine Basin a layer with higher salinity and AOU maxima related with pre-EMT-EMDW with an Adriatic Sea origin. In this basin, below 1500 dbar a more saline water and higher in oxygen pointed to EMDW with an Aegean Sea origin produced during the EMT, although more mixed compared to that found in the Ionian Sea. Here salinity and AOU were lower. The third EMDW is that found in the near-bottom layer of the Ionian Sea, it was produced after the EMT ($> 2001$) and has an Adriatic Sea origin (Hainbucher et al., 2013).

The upper panels in Figs. 5, 6 and 7 show the distribution of CO$_2$ species from the centre of the Ionian Sea to the Adriatic Pit. Here, below 3000 dbar, salinity increased in about 0.01 units (not shown), and TA as well (not resolved by the contour or the colour scale in Fig. 5) and DIC (Fig. 6) increased towards the bottom by about $4 \mu\text{mol kg}^{-1}$. Correspondingly, pH$_{25T}$ decreased by about 0.002 (note the greenish colour in the bottom of the upper panel in Fig. 7). The tracers distributions, CFC-12 and SF$_6$ (Stöven, 2011), also increased towards the bottom. These results point to a recent arrival of EMDW formed in the Adriatic Sea as commented previously.

v. Western Mediterranean Deep Water (WMDW): being fresher and colder than EMDW, the deep layers below 1500 dbar in the western MedSea during M84/3 were quite homogeneous in the physical (potential temperature and salinity) properties. A closer look showed two distinct WMDW varieties: a younger one, close to the bottom, as noted by the CFC-12 distribution (Stöven, 2011) and saltier than the older WMDW, immediately above 2000 dbar (Hainbucher et al., 2013). Bottom (younger) WMDW also had higher DIC values by about $7 \mu\text{mol kg}^{-1}$, lower pH$_{25T}$ values by about 0.004, and higher TA values by about $4 \mu\text{mol kg}^{-1}$, i.e. within the accuracy limits of the latter two variables (Figs. 5–7).
An outstanding feature in the vertical distribution of CO$_2$ variables was the pH$_{25T}$ minimum (Fig. 7) around 500 dbar in the Levantine and Ionian basins. In the Levantine basin this minimum was coincident with the oxygen minimum (Fig. 6 in Tanhua et al., 2013a) and a DIC maximum (Fig. 6). Whereas in the Ionian Sea the pH$_{25T}$ minimum/DIC maximum (500 dbar) was shallower than the oxygen minimum that was situated around 1000 dbars. All these extreme layers differ from the LIW layer, the salinity maximum was shallower in both basins, Levantine and Ionian. However, in the western basin, the LIW signal was coincident with the layer of highest remineralization as indicated by the DIC maxima and the oxygen/pH$_{25T}$ minimum.

During the M84/3 cruise we sampled the interior of the Tyrhenian Sea: in the upper layer low salinity (Tanhua et al., 2013a) and TA values (Fig. 5) pointed to AW. Around 400 dbar the maximum in salinity, TA, DIC and pH$_{25T}$ (Figs. 5, 6 and 7) were associated with LIW entering through the Sicily channel; Tyrhenian Deep Water (TDW) formed from the mixing of LIW and WMDW was detected below the TA maximum associated with LIW and bottom water with lower TA and pH$_{25T}$ and higher DIC values (Figs. 5, 6 and 7). This bottom water corresponds to WMDW that has started to cascade into the Tyrrhenian Sea from the western basin in 2011, also higher in tracer concentrations (Stöven and Tanhua, 2013).

As a reference for future studies and calculations in Sect. 4.1.2, Table 2 shows the mean values and standard deviation for salinity and different CO$_2$ species in the MedSea sub-basins and Atlantic waters west of the Strait of Gibraltar as sampled in 2011 during the Meteor M84/3 cruise by different depth intervals.

The direct comparison of our CO$_2$ data with previously published CO$_2$ data, either DIC, TA or pH, in order to identify and quantify trends is out of the scope of our manuscript. The available data for CO$_2$ variables in the MedSea is quite heterogeneous (see review in Álvarez, 2012, and below). A detail study about CO$_2$ temporal changes in the interior MedSea will be the aim of a forthcoming work. We do however make a rough comparison in order to detect possible biases in the historic dataset as compared to ours and vice versa. Qualitatively, our TA and DIC data in the western MedSea seems to be 5–10 and 40–70 µmol kg$^{-1}$ lower, respectively, than those shown in Millero et al. (1979) from a cruise in 1976. This result is counterintuitive due to the expected increase in DIC due to input of anthropogenic carbon ($C_{\text{ANT}}$). However, the data from Millero et al. (1979) originates before the use of certified reference materials (CRMs), so that the accuracy of these early data can be questioned. Our TA and DIC are however more comparable to those from the Almofront cruise in 1991 (Copin-Montegut, 1993). The data shown by Touratier and Goyet (2009) from the DYFAMED site in the Ligurian Sea below 1000 dbar, corresponding to WMDW, are lower than our WMDW values by about 5 µmol kg$^{-1}$ for both DIC and TA.

The reference cruise for the CO$_2$ system in the eastern and part of the western MedSea is the M51/2 cruise (R/V Meteor 2001) available at CDIAC; the TA data was presented in Schneider et al. (2007) and the DIC in Schneider et al. (2010). The TA data are comparable, but our DIC are roughly 10 µmol kg$^{-1}$ higher, which could be a real signal considering the 10 yr between the measurements and the uptake of anthropogenic carbon.

The VECTOR cruise in 2007 sampled 8 stations in the eastern and western MedSea for pH and TA, whereas DIC was calculated (Rivaro et al., 2010). Examining data from the western basin below 700 dbar, the TA data are comparable to each other whereas the DIC and pH values reported by Rivaro et al. (2010) are higher than ours by 10 µmol kg$^{-1}$ and by 0.01, respectively. In the eastern basin, our data below 700 dbar for pH and TA are comparable but our DIC seems 20 µmol kg$^{-1}$ higher. More recent CO$_2$ data are reported from the BOUM cruise in 2008 (Pujo-Pay et al., 2011; Touratier and Goyet, 2012); their DIC data seem higher than ours by about 10 µmol kg$^{-1}$ in both the eastern and western basins.

**Sardinia Sicily passage**

The vertical distributions of physical properties in the Sardinia Sicily passage, already analysed in the last decade of last century (Astraldi et al., 2002), indicate that the passage is a crucial chokepoint where almost all MedSea waters can be intercepted (see the review by Astraldi et al., 2002). The Sardinia Sicily transect, for its local character, was not included in the overall MedSea description done by Tanhua et al. (2013a). The main water masses crossing the passage have been identified according to their physical characteristics using potential temperature and salinity (Fig. 8a and b), AOU, (Fig. 8c) and CO$_2$ variables (Fig. 8d–f). On the eastern part (close to Sicily) only inflowing waters can be observed, from surface to bottom: (1) a fresh vein of AW (0–200 dbar), coming directly from Gibraltar through the Algerian sub-basin, was evident from the low salinity ($\approx 38.00$) (Fig. 8b) and supported by the lowest TA values (2525–2575 µmol kg$^{-1}$) (Fig. 8d); this inflowing AW was not in equilibrium with atmospheric oxygen (AOU between 20–40 µmol kg$^{-1}$, Fig. 8c), pH$_{25T}$ ($\approx 7.950$, Fig. 8f) and DIC values (2225–2285 µmol kg$^{-1}$, Fig. 8e) were the lowest observed along the Sect. 2) Below AW, the inflow of the saltiest water ($> 38.75$) in the transect, LIW, coming from the eastern MedSea, after crossing the Sicily Channel, the high AOU (50–60 µmol kg$^{-1}$) and DIC ($\approx 2300$ µmol kg$^{-1}$) values aside of the low pH$_{25T}$ ($< 7.940$) highlight the organic matter remineralization processes occurring within the LIW; TA values as high as 2605–2610 µmol kg$^{-1}$ were observed. (3) Below LIW, there was a thin layer of dense transitional Eastern Mediterranean Deep Water (tEMDW, i.e. the upper layer of the deep water filling the Eastern Mediterranean, which is able to cross the channel); tEMDW is cooler and...
fresher than LIW (<13.8 °C and <38.65), as well as for the higher DIC, TA (2318 and 2612 µmol kg⁻¹, respectively, Fig. 8d and e) and inorganic nutrient values (not shown) and the lower pH25T (<7.930, Fig. 8f); given its high density tEMDW cascades at greater depths to the interior of the Tyrrhenian Sea, contributing to the formation of the Tyrrhenian Deep Water, TDW, and provoking an intense mixing that dilutes its original physical properties (tEMDW is significantly warmer and saltier than TDW), in contrast the chemical properties of these two deep water masses remain very similar (Fig. 8).

On the western part (close to Sardinia) there are both inflowing and outflowing water masses, from surface to bottom: (1) close to Sardinia the inflowing AW has circulated through the Tyrrhenian Sea, becoming saltier (38.00 < salinity < 38.20, note the depth decrease towards Sardinia of the 38 isoline, Fig. 8b) and even less saturated in oxygen (AOU 35–40 µmol kg⁻¹, note the depth decrease towards Sardinia of the 20 isoline, Fig. 8c) than the inflowing AW on the Sicilian side; CO₂ variables are also slightly different (2550 < TA < 2575 µmol kg⁻¹; 2270 < DIC < 2285 µmol kg⁻¹; 7.940 < pH25T < 7.950, note the sloping isolines in Fig. 8d–f). (2) Below the outflowing AW, an outflowing LIW vein can be noted, which in turn, after its path through the whole Tyrrhenian Sea has become colder (∼14.0 °C) and less salty (<38.68) than the inflowing LIW on the eastern edge of the section, due to the mixing; this fact also lowers TA (∼2600 µmol kg⁻¹).

Table 2. Mean values and standard deviation for salinity and CO₂ species in the different MedSea sub-basins and west of the Strait of Gibraltar as sampled during the M84/3 cruise. Several vertical intervals are considered: surface (0–150 dbars in the East and 0–200 dbars in the West MedSea), intermediate (East: 150–500 and West: 200–500 dbars), deep (East: 500–2500 and West: 600–2000 dbars) and bottom (East: >2500 and West: >2000 dbars). IS stands for in situ values. Ar stands for aragonite and Ca for calcite.
Fig. 8. Vertical distributions across the Sardinia Sicily passage using EF11 data of (a) potential temperature (°C), (b) salinity, (c) Apparent Oxygen Utilisation (AOU, µmol kg⁻¹), (d) TA (µmol kg⁻¹) (e) DIC (µmol kg⁻¹) and (f) pH25T. Potential temperature, salinity and AOU are derived from CTD data, CTD-oxygen was calibrated with bottle data.

Fig, 8d) and increases DIC (∼2310 µmol kg⁻¹, Fig. 8e) while the pH25T (Fig. 8f) remains similar; outflowing LIW is also characterised by an absolute AOU maximum (65 < AOU < 73 µmol kg⁻¹, Fig. 8c) due to ageing. (3) In the central part of passage, over the plateau, there is an inflow of “old LIW” (Astraldi et al., 2002) (identified by its salinity maximum, but less salty than the outflowing LIW), also identified by higher AOU (55–70 µmol kg⁻¹, Fig. 8c) than the inflowing LIW (50–60 µmol kg⁻¹), while differences in CO₂ variables cannot be distinguished. (4) In deep layers (600–1900 dbars) the properties of LIW progressively become those of TDW, exiting the Tyrrenhian Sea, pH25T show the lowest values (< 7.930, Fig. 8f), AOU and DIC the highest ones (72 < AOU < 74 µmol kg⁻¹; 2315 < DIC < 2320 µmol kg⁻¹, Fig. 8c and e) and TA values (> 2605 µmol kg⁻¹, Fig. 8d) are not particularly characterising. (5) At the bottom and squeezed on the eastern wall in the deep canyon, there is an inflowing vein of Western Mediterranean Deep Water, which can be distinguished from the TDW because is fresher (< 38.52) and colder (< 13.4 °C), has slightly lower AOU (< 70 µmol kg⁻¹), higher pH25T values (> 7.940) but similar DIC (> 2318 µmol kg⁻¹) and TA (2605–2610 µmol kg⁻¹) values.

4.1.2 Derived variables

We used measured TA, DIC and inorganic nutrients data to calculate the degree of saturation (Ω) for calcite and aragonite using the preferred set of constants (see Sect. 3.2). The whole MedSea is supersaturated (Ω > 1) with respect to both calcite and aragonite throughout the whole water column. Figure 9 shows the vertical distribution of Ω, a clear longitudinal gradient is evident for surface, intermediate and deep waters with increasing values eastwards and a clear separation between the western, eastern MedSea basins and Atlantic waters to the west of the Strait of Gibraltar. The Ω values reported here compare well with those found by Millero et al. (1979) and Schneider et al. (2007). According to the classification by Kleypas et al. (1999) only the eastern basin in the upper 200 dbars presents adequate to optimal conditions for the development of coral reefs with Ω-Ar > 3.5.

Several projects, finished (e.g. EPOCA) or on going (e.g. BIOACID, MedSeA), try to decipher the combined effect of acidification and warming on the highly adapted MedSea
The absolute values of the buffer factors increased, 2014 www.ocean-sci.net/10/69/2014/
The DIC/TA ratio decreased eastwards in the MedSea; Introduction
References
Figures
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Fig. 9. Vertical distribution of the carbonate saturation state for calcite (circles) and aragonite (diamonds). Data from the EF11 cruise are shown as black circles (saturation of calcite) and diamonds (aragonite). The colour code stands for longitude.
calcareous and non-calcareous organisms. As a result of being too small and too complex in its circulation and biogeochemistry, the MedSea is usually not included in global projections models (e.g. Ricke et al., 2013) and a specific ocean-biogeochemistry model needs to be used. Here we provide a basin wide perspective for the CO2 chemistry in the MedSea as a basis for future studies and projections.

In order to quantify the ability of MedSea waters to resist changes in the CO2 system we calculated the buffer factors formulated by Egleston et al. (2010). The six factors proposed in their Table 1 quantify the sensitivity of aqueous CO2 [CO2] (γi), protons [H+] (βi) and the carbonate saturation state Ω (ωi) to changes in DIC and TA (subindex i) when the other parameter is kept constant. The calculations were performed with the output from CO2SYS at in situ conditions with the preferred set of constants and measured DIC and TA as inputs. The buffer factors have dimensions of mol kg⁻¹ × 10⁻³ of seawater (mmol kg⁻¹ hereinafter). See the formulation in Table 3 as several editing errors were detected in the original equations from Egleston et al. (2010).

The buffer factors γi express the fractional change in CO2 when DIC changes at constant TA (air-sea CO2 exchange) or when TA changes at constant DIC (additions of strong acid or base). The traditional Revelle factor, R, is directly calculated as \( R = \text{DIC}/\gamma_{\text{DIC}} \). The buffer factor \( \beta_{\text{TA}} \) expresses the fractional change in hydrogen ion concentration or activity when TA changes at constant DIC (addition of a strong base or acid). The factor \( \beta_{\text{DIC}} \) is identical to \(-\gamma_{\text{TA}}\). Since [Ca²⁺] and the solubility products of carbonate are invariant to DIC and TA changes, the buffer factors \( \omega_{\text{DIC}} \) account for the fractional change in \([\text{CO}_2^-]\) when DIC or TA changes when correspondingly TA or DIC is constant. A natural process just inducing strong acid/base additions is difficult to imagine, but a cargo disaster could be possible, unfortunately. A change in natural processes (primary production/mineralization of organic matter, formation/dissolution of carbonates, freshwater balance) usually affect both TA and DIC and therefore, the sensitivity of \([\text{CO}_2]\), pH and \([\text{CO}_2^-]\) to a change in those processes can be calculated using a combination of buffer factors (Egleston et al., 2010).

The values of \( \gamma_{\text{DIC}} \) and \( \beta_{\text{DIC}} \) are positive while \( \omega_{\text{DIC}} \) is negative because adding CO2 to seawater increases \([\text{CO}_2]\) and \([\text{H}^+]\) but decreases \([\text{CO}_2^-]\). Likewise, \( \gamma_{\text{TA}} \) and \( \beta_{\text{TA}} \) are negative while \( \omega_{\text{TA}} \) is positive because adding a strong base to seawater decreases \([\text{CO}_2]\) and \([\text{H}^+]\) but increases \([\text{CO}_2^-]\). Low buffer values imply low buffering capacity for a given change in DIC or TA. Minimum absolute values for the six buffer factors are found in waters with similar DIC and TA (DIC/TA ratio \( \approx 1 \)) and pH of about 7.5.

The six buffer factors, \( R \) and the DIC/TA ratio are shown in Fig. 10 and their mean values by sub-basin in Table 3. As the DIC/TA ratio approaches one the buffer capacities are reduced since \([\text{CO}_2]\) is more sensitive to temperature changes at this DIC/TA ratio. From the vertical distributions in Fig. 10 we can conclude that:

- The DIC/TA ratio decreased eastwards in the MedSea; being low compared to Atlantic waters. The highest ratios, i.e. higher \([\text{CO}_2]\) sensitivity to temperature, were associated with the Mediterranean Overflow, DIC/TA ratio \( > 0.9 \) (Fig. 10h).
- The absolute values of the buffer factors increased eastwards in the MedSea, therefore, the CO2 system in the eastern basin is less sensitive to changes in DIC and/or TA than the western basin. Atlantic waters, to the west of the Strait of Gibraltar, would be more sensitive than waters in the MedSea.
- The absolute buffering capacities distribute as \( \omega_{\text{DIC}} > \omega_{\text{TA}} > |\beta_{\text{TA}}| > (\beta_{\text{DIC}} = -\gamma_{\text{TA}}) > \gamma_{\text{DIC}} \). That is, the sensitivity of CO2 concentration to changes in DIC at constant TA (air-sea CO2 exchange) is the highest, followed by the change in pH due to the air-sea CO2 exchange (equivalent to the change in \( \text{CO}_2 \) due to a strong acid/base addition), then changes in pH due to strong acid/base additions. Being \( \omega_{\text{DIC}} \) and \( \omega_{\text{TA}} \) the highest buffer values means that MedSea waters are able to buffer changes in the saturation state due to additions of a strong acid or base and the air-sea CO2 exchange more easily than Atlantic waters.
Table 3. Mean values and standard deviation of several buffer factors in the different MedSea sub-basins and west of the Strait of Gibraltar as sampled during the M84/3 cruise. Same vertical intervals as in Table 2. The buffer factors have units of mmol kg$^{-1}$ and Revelle is dimensionless.

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<td>−0.20±0.01</td>
<td>−0.21±0.01</td>
<td>0.22±0.01</td>
<td>12.6±0.3</td>
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</tbody>
</table>

<table>
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<tr>
<th>BOTTOM</th>
<th>γDIC</th>
<th>βDIC</th>
<th>ωDIC</th>
<th>γTA</th>
<th>βTA</th>
<th>ωTA</th>
<th>Revelle</th>
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<tr>
<td>Aegean</td>
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<td>−0.30±0.01</td>
<td>−0.24±0.01</td>
<td>−0.26±0.01</td>
<td>0.29±0.01</td>
<td>9.4±0.1</td>
</tr>
<tr>
<td>Levantine</td>
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<td>−0.33±0.01</td>
<td>−0.27±0.01</td>
<td>−0.29±0.01</td>
<td>0.32±0.01</td>
<td>10.4±0.1</td>
</tr>
<tr>
<td>Ionian</td>
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<td>0.25±0.01</td>
<td>−0.31±0.01</td>
<td>−0.25±0.01</td>
<td>−0.27±0.01</td>
<td>0.29±0.01</td>
<td>11.0±0.1</td>
</tr>
<tr>
<td>Adriatic</td>
<td>0.21±0.01</td>
<td>0.25±0.01</td>
<td>−0.30±0.01</td>
<td>−0.25±0.01</td>
<td>−0.27±0.01</td>
<td>0.29±0.01</td>
<td>11.1±0.1</td>
</tr>
<tr>
<td>Atlantic</td>
<td>0.21±0.01</td>
<td>0.25±0.01</td>
<td>−0.30±0.01</td>
<td>−0.25±0.01</td>
<td>−0.27±0.01</td>
<td>0.29±0.01</td>
<td>11.1±0.1</td>
</tr>
</tbody>
</table>

γDIC = $\frac{DIC - TA}{TA}$
γTA = $\frac{TA - DIC \times S}{DIC \times S}$
βDIC = $\frac{DIC \times S - TA}{DIC \times S}$
βTA = $\frac{TA - DIC}{DIC}$
ωDIC = $\frac{DIC \times (2 \times [CO_2] + [HCO_3^-])}{TA}$
ωTA = $\frac{TA - DIC}{DIC}$

S = $[HCO_3^-] + 4 \times [CO_2^-] + \frac{[H^+] \times [H_2OH_4^+] - [H^+] + [OH^-]}{K_{ab}}$
P = $[HCO_3^-] + \frac{[H^+] \times [H_2OH_4^+] - [H^+] + [OH^-]}{K_{ab}}$

TA = $[HCO_3^-] + 2 \times [CO_2^-] + [CO_2]$
By examining the Revelle factor it is obvious that for a given change in CO$_2$ the relative change in DIC for MedSea waters is higher than in the adjacent Atlantic waters. That means that MedSea waters have the ability to store more anthropogenic carbon (C$_{\text{ANT}}$) than Atlantic waters. Within the MedSea the eastern basin is more prone to absorb more C$_{\text{ANT}}$ for a given CO$_2$ increase than the western basin.

The impact of the three main anthropogenic processes altering the surface CO$_2$ chemistry in MedSea waters is given in Table 4, which shows absolute changes in aqueous CO$_2$ ([CO$_2$]), pH and the aragonite saturation (Ω-Ar) from the initial surface values in Table 2; the different sub-basins are treated separately. A C$_{\text{ANT}}$ input through air-sea exchange (affecting DIC at constant TA) increasing DIC by 10 µmol kg$^{-1}$ will lead to a increase of [CO$_2$] from 0.56 to 0.78 µmol kg$^{-1}$, a pH decrease of around 0.016 units in MedSea waters, less than the pH decrease in Atlantic waters (0.020) for the same perturbation; the saturation state is reduced by about 0.1 units, but a reduction of at least 0.9 is required for deep waters to become under-saturated in aragonite. A hypothetical increase in calcification, resulting in a DIC increase of 10 µmol kg$^{-1}$ and double in TA, will compensate the pH and Ω decrease due to the input of C$_{\text{ANT}}$ and roughly half the corresponding increase in [CO$_2$]. Within the MedSea changes in [CO$_2$] and pH increase westwards being smaller than in Atlantic waters. An increase in salinity of 0.03 will cause minor changes in the CO$_2$ chemistry. The buffer factors here provided are useful to predict changes in the CO$_2$ chemistry as a function of DIC and/or TA changes due to different processes, natural and/or anthropogenic. However, it is relevant to relate the increase of C$_{\text{ANT}}$ in the ocean to the increasing partial pressure of CO$_2$ (pCO$_2$) in the atmosphere instead of using a DIC increase as reference. Table 5 shows the absolute changes in DIC, pH and Ω-Ar from initial states in Table 2 as a consequence of prescribed pCO$_2$ increases. Note that a pCO$_2$ increase of 25 µatm causes changes in DIC by about 10 µmol kg$^{-1}$, as already presented in Table 4. The main result in Table 5 is that as the Revelle factor in warm, salty and high TA waters is lower and the other buffer factors distribute as already presented in Table 4. The Revelle factor and DIC/TA ratio, both dimensionless.

### 4.2 Water mass characterisation

Here we present property-property plots for general physical and chemical variables, typical potential temperature-salinity, AOU-salinity and silicate concentration (Si$_{\text{II}}$)-salinity (Figs. 11 and 13) and the CO$_2$ variables, TA, DIC and
Table 4. Absolute associated changes in aqueous [CO$_2$] (µmol kg$^{-1}$), pH (no units) and aragonite saturation (no units) state due to several processes related to global change: a DIC increase of 10 µmol kg$^{-1}$ at constant TA (CO$_2$ air-sea exchange), a decrease in calcification of 10 µmol kg$^{-1}$ in DIC and double in TA, a salinity increase of 0.03. Surface values from Table 2 and 3 are taken to perform the calculations.

<table>
<thead>
<tr>
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<th>ΔDIC = +10 µmol kg$^{-1}$</th>
<th>ΔpH</th>
<th>ΔOmega</th>
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</thead>
<tbody>
<tr>
<td>Aegean</td>
<td>0.56</td>
<td>-0.016</td>
<td>-0.10</td>
</tr>
<tr>
<td>Levantine</td>
<td>0.57</td>
<td>-0.016</td>
<td>-0.10</td>
</tr>
<tr>
<td>Ionian</td>
<td>0.62</td>
<td>-0.017</td>
<td>-0.07</td>
</tr>
<tr>
<td>Adriatic</td>
<td>0.62</td>
<td>-0.016</td>
<td>-0.10</td>
</tr>
<tr>
<td>Tyrrenian</td>
<td>0.77</td>
<td>-0.018</td>
<td>-0.10</td>
</tr>
<tr>
<td>Western</td>
<td>0.76</td>
<td>-0.018</td>
<td>-0.10</td>
</tr>
<tr>
<td>Atlantic</td>
<td>0.78</td>
<td>-0.020</td>
<td>-0.10</td>
</tr>
</tbody>
</table>

Table 5. Absolute associated changes in DIC concentration, pH and aragonite saturation state due several partial pressure of CO$_2$ in the atmosphere increase. The initial conditions are the mean 2011 values (Table 2).

<table>
<thead>
<tr>
<th></th>
<th>ΔDIC in µmol kg$^{-1}$</th>
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<tr>
<td></td>
<td>25 µatm</td>
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<tr>
<td>Aegean</td>
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<td>Levantine</td>
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<td>12.9</td>
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<tr>
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<td>Tyrrenian</td>
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<td>Western</td>
<td>9.8</td>
</tr>
<tr>
<td>Atlantic</td>
<td>10.8</td>
</tr>
</tbody>
</table>

4.2.1 Aegean Sea (stations 287, 288 and 289)

Three stations during the M84/3 cruise were located in the Aegean Sea: 287 in the Chios Basin (central Aegean Sea), 289 in the eastern Cretan basin and 289 in the Kasos Strait (both in the southern Aegean Sea). In the upper 100 dbars high salinity identifies LSW (Levantine Surface Water) entering the Aegean Sea through the eastern Cretan arc straits (Fig. 11a). This LSW was also identified by the highest pH25T values measured in the M84/3 cruise (> 8.01, Fig. 12h) and also high TA and DIC values (∼2620 and ∼2280 µmol kg$^{-1}$, respectively, Fig. 12a and d). At intermediate levels, 100 to 250 dbars, CIW (Cretan Intermediate Water) presented the highest TA values for the M84/3 cruise (∼2645 µmol kg$^{-1}$, Fig. 12a and d). Deep layers below 2500 dbars were occupied by CDW (Cretan Deep Water), warm, salty and very dense (Fig. 11a), also identified by the distinct SiO$_2$-salinity relation (Fig. 11c), practically constant DIC (∼2320 µmol kg$^{-1}$) and TA (∼2630 µmol kg$^{-1}$) values vs. SiO$_2$ (Fig. 12b and e).
In the deep layers below 1000 dbars within the Levantine basin, the salinity minimum and then a pronounced temperature-salinity inversion (temperature and salinity increase) typically detected during the EMT (Roether et al., 2007) was still detected in this basin (Hainbucher et al., 2013) (red lines in Fig. S1). Water in the inversion is EMDW with an Aegean Sea origin (Aeg-EMDW) produced during the EMT.

Adriatic EMDW (Ad-EMDW) to the west of the East Mediterranean Ridge was differentiated from Aeg-EMDW mainly with AOU and SiO₂ (Fig. 11b and c), both being lower in the Ad-EMDW, which also contained higher transient tracer values (Stöven and Tanhua, 2013), indicating a more recent origin. It is difficult to separate them using TA, DIC and pH25T, both EMDW presented similar values, Ad-EMDW was slightly higher in DIC (around 5 μmol kg⁻¹) and TA (around 4 μmol kg⁻¹) and lower in pH25T (0.005 pH units) than Aeg-EMDW (Fig. 12c, f and i) (blue and black lines in Fig. S1).

The salinity minimum around 1000 dbars and below the temperature-salinity inversion previously commented provoke a sort of hook in the relationship between DIC – TA – pH25T vs. AOU or SiO₂ (red lines in Fig. S1) for deep waters in this basin. DIC and TA decrease with decreasing SiO₂ or AOU until their corresponding maxima (salinity minimum) and then towards the bottom DIC and TA increase with decreasing AOU (increasing salinity). pH is the opposite, it decreases with AOU increasing until the AOU maximum (salinity minimum) and then pH increases and AOU decreases (Fig. S1).

### 4.2.3 Ionian Sea (from station 299 to 309, 314 and 315)

Two upper water masses could be differentiated in the Ionian Sea, AW (< 17 °C, < 38.9) and LSW (> 16 °C, > 38.8) (Fig. 11a). During M84/3 LSW in the Ionian Sea was found in the stations to the western part of the Hellenic Trench (Sts. 299, 300, 301) (Fig. 1). Below, two intermediate water masses, LIW and CIW could be distinguished. LIW with temperature and salinity maxima around 200 dbars was also associated with a TA and DIC maximum (Fig. 12a and d). The mixing between LIW and AW produced an exponential relationship between TA and DIC vs. salinity (Fig. 12a and d). Deviations from this relationship towards lower TA and DIC values corresponded to LSW in this basin. The other intermediate water detected, CIW, mixed with LIW south of Crete, in the eastern basin was found in stations 299, 300 and 301 to the west of the Antikithera strait. This water mass with also very high salinity, presented an even higher TA than LIW (Fig. 12a, b and c). For example in the case of the TA vs. SiO₂ plot (Fig. 12b) CIW was detected as a deviation towards higher values from the typical TA – SiO₂ relationship. CIW was also seen as a hump between 0 to 30 μmol kg⁻¹ AOU over the LIW data (Fig. 12c).

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**Fig. 11.** Property-property plots for the M84/3 data in the eastern basin: Aegean Sea (green dots), Levantine basin (magenta), Ionian Sea (blue) and Adriatic Sea (yellow). (a) Potential temperature (°C) vs. salinity, (b) Apparent Oxygen Utilisation (AOU in μmol kg⁻¹) vs. salinity and (c) silicate concentration (SiO₂ in μmol kg⁻¹) vs. salinity. The station positions are shown in Fig. 1.

**4.2.2 Levantine Basin (from station 290 to 298)**

In the upper water column, < 100 dbars, AW and LSW could be distinguished using salinity (Fig. 11a), LSW presented salinity > 38.9 and AW from 38.58 to 38.9 (Fig. 11a). TA, DIC and pH25T values much higher in LSW could also differentiate them (Fig. 12a, d and g). Interestingly, LSW presented pH25T values higher, and AW lower than 8.01 for waters with AOU < 0 (Fig. 12i).

At intermediate depths, 100–400 dbars, the main water mass was LIW (Levantine Intermediate Water). In addition to the typical scorpion tail relation between salinity and potential temperature, LIW was recognized by SiO₂ minima associated with a salinity maximum (Fig. 11c) and high TA (≈ 2630 μmol kg⁻¹, Fig. 12a) values.
Fig. 12. Property-property plots for the M84/3 data in the eastern basin: Aegean Sea (green dots), Levantine basin (magenta), Ionian Sea (blue) and Adriatic Sea (yellow). First row: alkalinity (TA in µmol kg\(^{-1}\)) vs. (a) salinity, (b) silicate concentration (SiO\(_2\) in µmol kg\(^{-1}\)) and (e) AOU (µmol kg\(^{-1}\)). Second row: Dissolved Inorganic Carbon (DIC in µmol kg\(^{-1}\)) vs. (d) salinity, (e) SiO\(_2\) and (f) AOU. Third row: pH\(_{25T}\) vs. (g) salinity, (h) SiO\(_2\) and (i) AOU. The station positions are shown in Fig. 1.

In the Ionian Sea, Post-EMT-EMDW occupied the layer below 1500 dbars, it presented lower AOU and SiO\(_2\) values than EMT Aeg-EMDW (Fig. 11b and c) in the Levantine basin. The recent replenishment from the Adriatic Sea, 2001 onwards, was detected as a temperature-salinity increase in bottom waters of the central Ionian Sea (Rubino and Hainbucher, 2007) (green and cyan lines in Fig. S2). These younger bottom waters in the Ionian Sea explain the DIC increase (Fig. 6), AOU and pH\(_{25T}\) (Fig. 7) decrease towards the bottom of the Ionian Sea (compare red line with the blue and cyan lines in Fig. S2).

The hook structure in the relationships DIC – TA – pH vs. AOU for deep waters in the Levantine basin (Fig. S1) was here eroded as is the temperature-salinity inversion associated with EMT Aeg-EMDW. The AOU maximum in the Ionian Sea was no longer as clearly associated with the salinity minimum as in the Levantine basin. Below 500 dbars DIC and TA decrease with increasing AOU, from the AOU maximum TA practically stays constant and DIC increases. The very recent input of bottom Ad-EMDW is noticed in the pH vs. AOU relationship, from 500 dbars pH decreases with AOU increasing, from the AOU maximum pH increases (AOU decreases) but close to the bottom both pH and AOU decrease (Fig. S2).

4.2.4 Adriatic Sea (stations 310 and 313)

Only two stations were sampled in the Adriatic Sea for CO\(_2\) variables, one in the Adriatic Pit (313) and one to the east of the Otranto Strait (310). One of the densest deep waters during the M84/3 cruise was sampled in the Adriatic Pit at 1100 dbars (Fig. 11a), this water remains confined as the Otranto Strait is only 800 m deep. Waters in the Adriatic presented high TA values despite a moderate salinity (Fig. 12a), they were out of the general linear relationship between TA and salinity due to the high TA of the rivers discharging in the area (Luchetta et al., 2010; Cantoni et al., 2012; Copin-Montegut, 1993). The relationship TA vs. SiO\(_2\) was also different for Adriatic waters (Fig. 12b), being lower than the general trend in the eastern MedSea. Additionally DIC and pH in the Adriatic had a distinct signal, high values for DIC and low values for pH in waters with salinity lower than 38.7 (Fig. 12d and g).
Fig. 13. Property-property plots for the M84/3 data in the western basin: Tyrrhenian Sea (green dots), Western basin (blue) and Strait of Gibraltar (magenta). Light black dots correspond to Ionian Sea data. (a) Potential temperature (°C) vs. salinity, (b) Apparent Oxygen Utilisation (AOU in µmol kg\(^{-1}\)) vs. salinity and (c) silicate concentration (SiO\(_2\) in µmol kg\(^{-1}\)) vs. salinity. The station positions are shown in Fig. 1.

The current overflow from the Adriatic stood out in the pH – AOU plot (Figs. 14i and 7, upper panel): pH data for AOU values between 40 and 45 µmol kg\(^{-1}\) in stations 309 and 310 located in the Otranto Strait, deviated towards 7.94 pH25T values, corresponding to waters around 500 dbars in the Adriatic Pit.

4.2.5 Tyrrhenian Sea (from station 316 to 321)

This basin acts as a connector between the eastern and the western MedSea, here intermediate waters entering from the east mix with deep waters entering from the west (e.g. Hopkins, 1988; Astraldi et al., 1996). This fact was noticed in the property-property plots shown in Fig. 13 were the points corresponding to the Tyrrhenian Sea clearly connected the western and eastern MedSea waters.

In the surface layer, AW with salinity lower than 38 was detected in all the Tyrrhenian stations except 319 in the centre of the basin (Fig. 13a). The freshest AW was seen in station 316, the nearest to Sicily, here AW comes directly from Gibraltar through the Algerian sub-basin. AW was also distinguished by low AOU and SiO\(_2\) values close to zero (Fig. 13b and c). The minimum TA and DIC associated with AW are 2466 and 2178 µmol kg\(^{-1}\), respectively (Fig. 14a and d). In the centre of the Tyrrhenian Sea the upper 52 dbars were occupied by probably modified AW with 38.5 > salinity > 38 affected by evaporation or mixed with the salty intermediate water entering through the Strait of Sicily. This water also presented a different relationship between salinity and pH25T compared to unmix AW (Fig. 14g).

Maximum salinity values in the intermediate layer between 250 and 500 dbars pointed to LIW entering from the eastern basin (Fig. 13a). As LIW flows into the Tyrrhenian Sea remineralization processes increase its AOU, DIC and lower pH25T, with extreme values found in St. 319 (cyan line in Fig. S3) in the interior of this basin and at St. 321 (magenta line in Fig. S3) in the Sardinia Channel where aged LIW flows out of the Tyrrhenian Sea into the western basin.

At St. 316 (red line in Fig. S3), the nearest one to Sicily in the Sardinia Sicily passage, the intermediate layer was occupied by a fresher LIW (38.66–38.68) with higher AOU and DIC (lower pH25T and TA) values. It might be LIW mixed with AW returning to the western basin although this is an area where LIW is supposed to enter the Tyrrhenian Sea.

Deep layers in the Tyrrhenian are occupied by TDW with a typical linear relationship in the temperature-salinity plot (Fig. 13a). No hook in the temperature-salinity diagram to lower salinities was detected in bottom waters of the Tyrrhenian Sea, this hook would show the input of recently formed WMDW as detected in the bottom of the western basin. In the Tyrrhenian Sea bottom WMDW stands out from TDW by the sharp decrease in AOU and pH25T (Fig. 7) and increase in DIC (Fig. 6) for example with salinity (red dots in Fig. S4).

4.2.6 Western basin (from station 322 to 336)

The upper layer in this basin was occupied by AW very fresh (≪38) and practically 100% saturated in oxygen (AOU ≈ 0 µmol kg\(^{-1}\)) and with no SiO\(_2\) (Fig. 13b and c). In the intermediate layer the salinity maximum associated with LIW evolves from 38.67 at St. 322 to 38.51 at St. 336. There was a marked reduction in salinity (and TA) when LIW entered the western basin passing from 38.71 at St. 321 in the Sardinia Channel, 38.67 at St. 322 (right west of the channel) and then 38.51 at St. 323 in the western basin. The change in the other chemical variables, SiO\(_2\), AOU, DIC or pH25T was less acute.
When in the western basin, LIW was associated with a maximum in AOU (Fig. 13b), DIC (Fig. 14d) and minimum in pH25T (Fig. 14g), so it was the layer of maximum remineralization of organic matter, contrary to the eastern basin where this layer was right below LIW.

The near bottom layer in the western basin was occupied by a recently formed WMDW with higher salinity, so the temperature-salinity plots present a sort of hook structure (Hainbucher et al., 2013). This new WMDW presented higher DIC and lower AOU than the older WMDW (Figs. 14f and S5).

4.2.7 Atlantic waters (from station 337 to 344)

Due to the dramatic scale change, Figs. 13 and 14 show only the data up to St. 338 (corresponding to the Strait of Gibraltar). All the samples collected to the west of this Strait are shown in Fig. S6.

At the Strait of Gibraltar (St. 337 and 338) the intermediate LIW outflowing presented a salinity of 38.48 (TA $\approx$ 2584 µmol kg$^{-1}$, DIC $\approx$ 2331 µmol kg$^{-1}$, pH25T $\approx$ 7.89, AOU $\approx$ 75.3 µmol kg$^{-1}$) (Figs. 14 and S6). The outflow was only detected at stations 342 and 344 (pink lines in Fig. S6) in the Portuguese continental slope, with a salinity of 36.50 (TA $\approx$ 2409 µmol kg$^{-1}$, DIC $\approx$ 2193 µmol kg$^{-1}$, pH25T $\approx$ 7.86, AOU $\approx$ 58.2 µmol kg$^{-1}$). In the centre of the Gulf of Cádiz (339 and 340, black and green lines respectively in Fig. S6), the outflow was mixed with the upper low salinity central waters and presented a salinity of 36.20 (TA $\approx$ 2392 µmol kg$^{-1}$, DIC $\approx$ 2202 µmol kg$^{-1}$, pH25T $\approx$ 7.79, AOU $\approx$ 93 µmol kg$^{-1}$).

5 Summary

Two coordinated cruises sampled the Mediterranean Sea for CO$_2$ variables (TA, DIC and pH) in 2011: the RV Urania EF11 (Sardinia Sicily passage) and the RV Meteor M84/3 (East-West full-length transect).

The over-determined and high-quality CO$_2$ data collected in the M84/3 cruise allowed performing the first internal consistency analysis for CO$_2$ data in the peculiar warm, salty and low DIC/TA ratio MedSea waters. Using the $^\circ$MATLAB version for the consensus CO2SYS programme the preferred set of constants is option 4 [CO$_2$ constants from Mehrbach]
et al. (1973) refitted by Dickson and Millero (1987) along with the sulphate constant from Dickson (1990) and the parameterization of borate from Uppström (1974). Using this combination our pH, TA and DIC measurements from the M84/3 cruise are internally consistent to −0.0003 ± 0.005 for pH, 0.1 ± 3.3 µmol kg⁻¹ for TA and −0.1 ± 3 µmol kg⁻¹ for DIC. Despite this encouraging result further studies are needed including a fourth variable: pCO₂ in order to verify the internal consistency of K₁ only a reliable K₁ is needed to calculate pCO₂ from the input of pH – TA or pH – DIC and [CO₃²⁻] in order to revise the K₂ parameterization in low DIC/TA ratio waters where the CO₂ system is more sensitive to K₂. Using the combination of pH – TA – DIC only a reliable K₂ is needed in order to calculate bicarbonate, carbonate and aqueous CO₂ (Park, 1969). Particularly comparing measured (Byrne and Yao, 2008) and calculated [CO₃²⁻] could be helpful to revise K₂.

The Meteor M84/3 data allowed a post-EMT, full length and depth description of CO₂ variables in the MedSea. This study aims to be base-line for future data and model studies dealing with the effects of natural (climate) and human (warming, acidification, etc.) driven changes in the CO₂ species and ancillary variables associated.

As a general view, the eastern MedSea presents higher TA (2560–2644 µmol kg⁻¹) and pH25T (7.935–8.032) and less variable DIC (2247–2331 µmol kg⁻¹) than the western basin with TA (2388–2608 µmol kg⁻¹), pH25T (7.861–7.988) and DIC (2110–2336 µmol kg⁻¹). The saturation for calcite and aragonite in both basins is well above 1, for the eastern basin Ω-Ca vary 2.5–5.78 and Ω-Ar 1.73–3.80 and for the western basin Ω-Ca 2.73–5.25 and Ω-Ar 1.86–3.44.

The CO₂ chemistry in the warm, salty and high TA waters in the MedSea compared to the Atlantic is in general less sensitive to temperature changes because of the low DIC/TA ratio. MedSea waters are more resistant to direct changes in DIC and/or TA due to natural and/or anthropogenic processes, since the calculated buffer factors are higher in MedSea waters, showing an increasing westwards gradient. The CO₂ chemistry in the western basin would be more sensitive than in the eastern basin.

However, from the point of view of an increasing pCO₂ in the atmosphere, that increases indirectly DIC and keeps TA constant in the ocean, waters in the MedSea have a lower Revelle factor than in the Atlantic, and therefore are able to store more DIC for a given pCO₂ increase than in the Atlantic but the changes in pH (Ω) would be lower (higher) than in the Atlantic.

As a summary the water masses encountered according to depth are:

i. Levantine Surface Water (LSW): warm and salty (>38.9) surface water (upper 100 dbars) formed by intensive heating and evaporation in the Levantine basin, found during the M84/3 cruise in this basin and the Aegean and Ionian Seas. LSW was characterised by high TA (∼2610 µmol kg⁻¹) and DIC (∼2270 µmol kg⁻¹). In April 2011 LSW presented pH25T > 8.01 associated with AOU < 0 pointing to active photosynthetic activity.

ii. Atlantic Water (AW): it is the upper part (<125 dbars) of the open estuarine cell in the MedSea, enters through the Strait of Gibraltar and flows to the Levantine basin while suffering evaporation and heating. The main CO₂ variable showing the presence of AW is TA, which evolved from very low values in the Strait of Gibraltar (∼2390 µmol kg⁻¹) to high values (∼2610 µmol kg⁻¹) in the Levantine basin. Evaporation processes and mixing with the salty Levantine Intermediate Water (LIW) induced a practically linear relation between TA and salinity. Another linear relationship was found between pH25T and AOU for the western MedSea in April 2011 pointing to a biological modulation of pH.

iii. Levantine Intermediate Water (LIW): main intermediate water mass in the MedSea (125–500 dbars), is the lower part of the open estuarine cell in the MedSea. Typically LIW is identified by a scorpion tail in the temperature-salinity diagram associated with a salinity maximum. The main CO₂ variable identifying LIW in any MedSea sub-basin was an intermediate TA maximum evolving from ∼2620 µmol kg⁻¹ in the Levantine and Ionian basins to ∼2600 µmol kg⁻¹ in the Tyrrhenian Sea, ∼2590 µmol kg⁻¹ in the western basin, and ∼2584 µmol kg⁻¹ overflowing in the Strait of Gibraltar, and finally ∼2409 µmol kg⁻¹ in the Mediterranean Water located in the Gulf of Cadiz, that then circulates into the North Atlantic.

In the Levantine and Ionian basins LIW was found shallower (∼250 dbars) than in the western basin (∼450 dbars). In the eastern MedSea LIW was situated above the layer of maximum organic matter mineralization while in the western MedSea they coincided and therefore was associated with a maximum in DIC and AOU and minimum in pH.

iv. Cretan Intermediate Water (CIW): formed in the Cretan Sea, this intermediate (250 dbars) water was characterised by even higher TA (2630–2645 µmol kg⁻¹) than LIW and detected in the Aegean Sea and the Ionian Sea west of the Antikithera strait.

v. Cretan Deep Water (CDW): produced by convection in the Cretan Sea, was clearly identified apart of the high density, by a distinct relation between SiO₂ – salinity, high constant TA (2630 µmol kg⁻¹) and DIC (2320 µmol kg⁻¹) values with varying SiO₂ (2–4 µmol kg⁻¹). This water remained confined in the Aegean Sea.
vi. Pre-EMT Eastern Mediterranean Deep Water (EMDW): this water mass, of an Adriatic Sea origin, was still detected in the Levantine basin by a salinity minimum around 1000 dbars, it presented a minimum in tracer values as CFC-12 or SF₆, but also a minimum in pH because was the layer of organic matter remineralization as also indicated by the maximum in AOU and inorganic nutrients.

vii. EMT-EMDW: below the salinity minimum in the Levantine basin temperature and salinity increase producing a hook in the temperature-salinity diagram which is associated with the input of EMDW produced in the Aegean Sea during the EMT. Due to a more recent origin it presents lower AOU, slightly higher DIC and pH values than pre-EMT EMDW.

viii. Post-EMT-EMDW: bottom layers in the Hellenic Trench (Levantine basin) and deep layers (> 1000 dbars) in the Ionian Sea are occupied by EMDW with an Adriatic origin produced after the EMT. The hook in the temperature-salinity diagram was no longer detected and a less salty post-EMT-EMDW occupied the deep layer in the Ionian Sea. This water was also identified by lower AOU and pH values compared to the salter EMT-EMDW.

Bottom layers in the Ionian Sea, below about 3000 dbars, were occupied by recent overflow waters produced in the Adriatic Sea after 2001. It presented a clear signature in DIC (> 2310 µmol kg⁻¹), AOU (< 55 µmol kg⁻¹) and pH (< 7.967).

ix. Waters in the Adriatic Sea: surface waters in this basin presented a distinct TA vs. salinity relationship with values around 2610 µmol kg⁻¹ for salinity < 38.7, higher than the general TA–salinity relation mainly due to the high TA input through Adriatic rivers. The overflow from the Adriatic Sea was mainly detected in the pH vs. AOU relationship, for AOU values between 40–45 µmol kg⁻¹ pH deviated to 7.94, those are waters from around 500 dbars in the Adriatic Pit spilling into the Ionian Sea.

x. Waters in the Tyrrenhenian Sea: the property-property plots showed that this basin is a connector between the eastern and western basins, where LIW mixes with deep waters from the west producing Tyrrenhenian Deep Water (TDW). During 2011 no recently formed Western Mediterranean Deep Water (WMDW) was detected in the bottom of the Tyrrenhenian Sea.

The water mass exchange between the eastern and western MedSea is highlighted using data from the Urania EF11 cruise in the Sardinia Sicily passage: AW mainly enters the Tyrrenhenian on the eastern part of the passage, close to Sicily, and exits this basin on the Sardinian side, becoming saltier and also higher in TA; the same circulation scheme is done by LIW, which becomes colder and saltier after circulating in the Tyrrenh- nian, TA is decreased and DIC increased.

xi. Western Mediterranean Deep Water (WMDW): this water occupied layers below 1000 dbars in the western basin of the MedSea, with respect to CO₂ variables it presented DIC (2318–2320 µmol kg⁻¹) higher and TA (2584–2588 µmol kg⁻¹) and pH25T (7.90–7.91) lower than EMDW. In the bottom layer a recent input, after 2005, of WMDW was detected with higher salinity and temperature, while DIC (AOU) clearly increased (decreased) towards the bottom.

xii. In the Strait of Gibraltar LIW presented in 2011 a salinity of 38.48 and TA ≈ 2584 µmol kg⁻¹, DIC ≈ 2331 µmol kg⁻¹, pH25T ≈ 7.89, AOU ≈ 75.3 µmol kg⁻¹; once mixed with central waters in the Gulf of Cadiz, Mediterrane ane Water sampled in the Portuguese slope was detected with a salinity of 36.50, TA ≈ 2409 µmol kg⁻¹, DIC ≈ 2193 µmol kg⁻¹, pH25T ≈ 7.86, AOU ≈ 58.2 µmol kg⁻¹.

The CO₂ and ancillary data collected in 2011 highlight the need for a sustained programme monitoring the temporal evolution of water masses in the MedSea. This small but relevant marginal sea is a perfect laboratory basin to detect the effect of natural and anthropogenic driven changes in the physics and CO₂ chemistry. Special emphasis should be given to CO₂ variables as MedSea waters are especially sensitive to pCO₂ increases in the atmosphere: they are prone to absorb more DIC for given pCO₂ increase, this DIC increase in the surface is rapidly transported to depth with the overturning circulation, and therefore pH and carbonate saturation conditions are altered, with consequences still unknown for the well adapted organisms.

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References


Roether, W., Klein, B., Manca, B. B., Theocharis, A., and Kioroglou, S.: Transient Eastern Mediterranean deep waters in


